

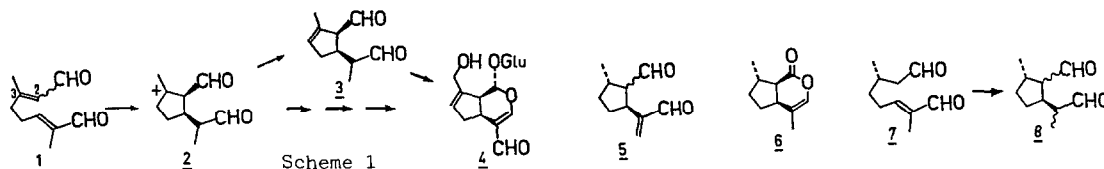
THE BASE-CATALYZED CYCLIZATION OF 10-OXOCITRAL
 SYNTHESIS OF CHRYSOMELIDIAL AND DEHYDROIROIDIAL

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Base treatment of 10-oxocitral (1) gives chrysolmelidial (9) and dehydroiridodial (10), supporting the intermediacy of 1 in the biosynthesis of some iridoid glucosides. Cannizzaro reaction of 9 and 10 affords regioselectively a new iridolactone, 1,2-dehydroisoiridomirmecin (13).

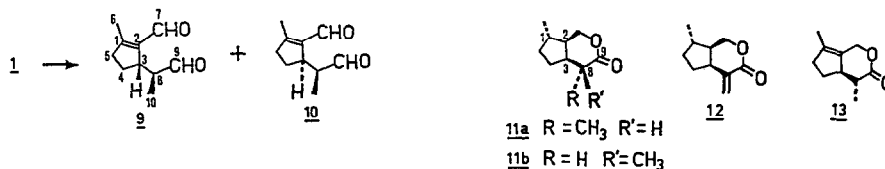
The intermediacy of 10-oxocitral (1) and iridodial cation (2) in the biosynthesis of iridoid glucosides of *Gardenia jasminoides* F. has recently been discussed¹ in this Journal (Scheme 1).



The biological cyclization from a diolefinic intermediate to the iridane skeleton is not a general pathway. For instance, in the case of cyclopentanoid metabolites, such as dolichodial² (5) or nepetalactone³ (6), structurally much simpler with respect to iridoid glucosides and indole alkaloids, 2,3-saturated acyclic precursors, such as 7, are involved in this step.

In the laboratory, the synthesis of iridodial (8) from 7 was realized⁴ by R. Robinson in 1959, the ring formation being a normal Michael reaction. In view of this chemical result and of the biosynthetic observations concerning 10-oxocitral, we attempted to carry out the cyclization of 1, to check if a cyclopentane structure, such as dehydroiridodial (3), could be obtained.

Acid-catalyzed reactions on 2E-1⁵ were quite unsuccessful, in any case affording only tars. Among the different bases⁶ and conditions we tried, just with 0.01 N NaOH in MeOH-H₂O (98:2) (0.3 Molar, 10' at r.t.) was a smooth conversion achieved. The isolated (82% yield) product was a 1.2:1 (25 m gc capillary column of SE 54) mixture of chrysolmelidial (9) (3R,8R and 3S,8S) and its diastereomer, dehydroiridodial (10) (3S,8R and 3R,8S), which are olefinic isomers of 3. This last compound was not found (gc, nmr) in the recovered material.



Structures **9** and **10** have been determined by spectral (gc-ms, nmr) comparison^{7,8} and chemical reduction⁸ (LiAlH_4) into the corresponding known⁸ isodehydroiridodiol and dehydroiridodiol, respectively. As natural compounds, they have been found in certain chrysomelid beetles^{7,9} (**9**, **10** and/or their enantiomers) and in the cat-attracting plant *Actinidia polygama* Miq.¹⁰ (**10**).

The feasibility of a cyclization process from a diolefinic dialdehyde to an iridane skeleton is thus proved. The conversion has been only achieved in a base-catalyzed reaction, in which the formation of the hypothesized cation species **2** is not expected. However, this or an equivalent intermediate is rather likely to be formed in the biological enzyme-mediated transformation, through an intramolecular general acid catalysis¹¹.

Lactones with the carbonyl group at C_9 , isoiridomirmecin (**11a**) and allodolicholactone (**12**), have been regioselectively obtained on subjecting iridodial (**8**)^{4,12} and dolichodial (**5**)¹³, respectively, to the Cannizzaro reaction conditions. By using this procedure¹³ on the isolated reaction mixture (1.2:1) of chrysomelidial (**9**) and dehydroiridodial (**10**), a new dehydroiridolactone (**13**) was obtained (71% yield), again regioselectively, bp 106–8°C/0.2 mmHg; m/e (%): 166 (M^+ , 13), 122 (51), 107 (100), 91 (56), 79 (56); ir (film), cm^{-1} : 1725, 1645; $^1\text{Hnmr}$ (CDCl_3), δ : 1.26 (3H, d (J 6.5 Hz), $\text{CH}_3\text{-C}$), 1.67 (3H, br s, $\text{CH}_3\text{-C=}$), 2.0–2.8 (6H) and 4.98 (2H, m, $\text{OCH}_2\text{-C}$).

The stereochemistry (3R,8R and 3S,8S) of **13** was established by reduction (LiAlH_4 , 85% yield) into the known⁸ isodehydroiridodiol. The formation of only one diastereomer (**13**), of the isoiridomirmecin (**11a**) type, from both **9** and **10** is easily explained with reference to the observed¹⁴ base-promoted transformation of iridomirmecin (**11b**) into isoiridomirmecin (**11a**).

A regioselectivity which in any case yields lactones (**11a**, **12** and **13**) with the carbonyl group at C_9 , starting from both one saturated (**8**) and two isomeric unsaturated aldehydes having the olefin conjugation at C_9 (**5**) or at C_7 (**9** and **10**) carbonyl function, is unexpected; it means that conjugation does not influence the hydride donor or acceptor ability of the two aldehyde functions in the Cannizzaro reaction intermediate¹⁵. Steric factors thus have priority over electronic ones, and one explanation may be that the formation of a tetrahedral gem-diol group in these iridoid substrates causes a more severe steric constraint at C_7 than at C_9 position.

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References: 1) S. Uesato, S. Ueda, K. Kobayashi, M. Miyauchi and H. Inouye, *Tetrahedron Lett.*, **25**, 573 (1984). 2) F. Bellesia, U.M. Pagnoni, A. Pinetti and R. Trave, *Phytochemistry*, **22**, 2197 (1983). 3) F. Bellesia, R. Grandi, U.M. Pagnoni, A. Pinetti and R. Trave, *Phytochemistry*, **23**, 83 (1984). 4) K.J. Clark, G.I. Fray, R.H. Jaeger and R. Robinson, *Tetrahedron*, **6**, 217 (1959). 5) J. Balsevich, *Can. J. Chem.*, **61**, 1053 (1983). 6) A.F. Thomas and R. Guntz-Dubini, *Helv. Chim. Acta*, **59**, 2261 (1976). 7) J. Meinwald and T.H. Jones, *J. Am. Chem. Soc.*, **100**, 1883 (1978). 8) T. Sakai, K. Nakajima, K. Yoshihara, T. Sakan and S. Isoe, *Tetrahedron*, **36**, 3115 (1980). 9) J.M. Pasteels, J.C. Braekman, D. Daloz and R. Ottinger, *Tetrahedron*, **38**, 1891 (1982). 10) K. Yoshihara, T. Sakai and T. Sakan, *Chemistry Letters*, 433 (1978). 11) *Comprehensive Org. Chem.*, eds D.H.R. Barton and W.D. Ollis, Pergamon, Oxford, 1979, Vol 4, p. 404. 12) G.W.K. Cavill and D.L. Ford, *Aust. J. Chem.*, **13**, 296 (1960). 13) U.M. Pagnoni, A. Pinetti, R. Trave and L. Garanti, *Aust. J. Chem.*, **29**, 1375 (1976). 14) R. Fusco, R. Trave and A. Vercellone, *Chim. Ind.*, **37**, 251 (1955); G.W.K. Cavill and H.D. Locksley, *Aust. J. Chem.*, **10**, 352 (1957). 15) C.G. Swain, A.L. Powell, W.A. Sheppard and C.R. Morgan, *J. Am. Chem. Soc.*, **101**, 3576 (1979).

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